#### **Unilateral NMR of Activated Carbon**

Stuart Brewer<sup>2</sup>, Hans Adriaensen<sup>1</sup>, Martin Bencsik<sup>1</sup>, Glen McHale<sup>1</sup> and Martin W Smith<sup>2</sup>

[1]: Nottingham Trent University (NTU), UK

dstl

[2]: Defence Science and Technology Laboratory (DSTL), UK

### **Background- Unilateral NMR**





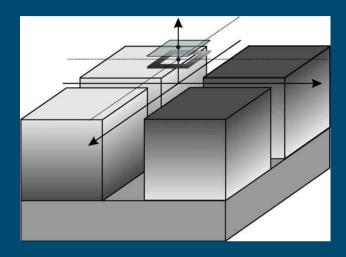
- The measurements are done on an object placed on top of the NMR Mouse®.
- The NMR Mouse® is designed to project a sensitive area outside the instrument that can be probed.
- Therefore with this design, experiments are performed outside rather than within the instrument.
- The polarising magnets assembly and the RF coil has the shape of a box (13 x 11 x 10 cm<sup>3</sup>).

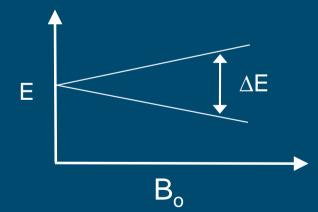
[1]: D.G. Rata, F. Casanova, J. Perlo, D.E. Demco, B. Blümich \*, JOURNAL OF MAGNETIC RESONANCE, 180 (2006) 229–235





#### Suitability of the NMR Mouse®





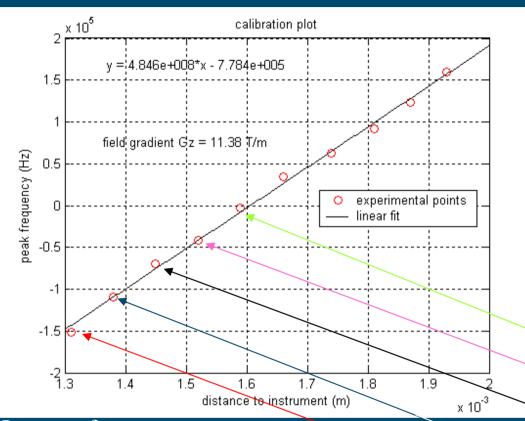
 <sup>1</sup> J.Perlo, JOURNAL OF MAGNETIC RESONANCE 176 (1): 64-70 SEP (2005)
<sup>2</sup> ACT, Aachen, Germany.

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- The magnet geometry generates a flat sensitive volume of an adjustable size of approximately 1.5 cm<sup>2</sup> and 0.6 mm thick at 2-10mm away from the instrument.
- A radio frequency field coil excites and detects the proton nuclear spins.
  - The measurement essentially yields; the proton density
  - the relaxation rate of the nuclear spins.



### **Field Gradient Effect**



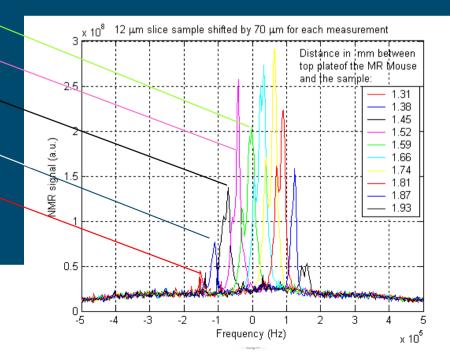
Larmor frequency:

 $ω = γB_0$   $2π\delta f = γG_z\delta$   $G_z = (2π/2.67522 \times 10^8)(4.846 \times 10^8)$  $G_z = 11.38 \text{ T.m}^{-1}$ 

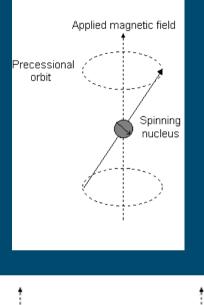
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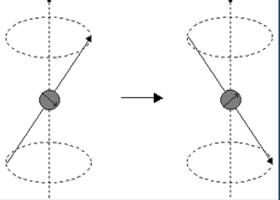
The z-coordinate of a 12 µm thin film of sandwiched oil between two microscopes slides was incrementally changed by interleaving a gradually increasing stack of 70 µm thick tracing paper sheets.

Excellent linearity is found resulting in an approx Gaussian profile.



### Information from Relaxation Rates



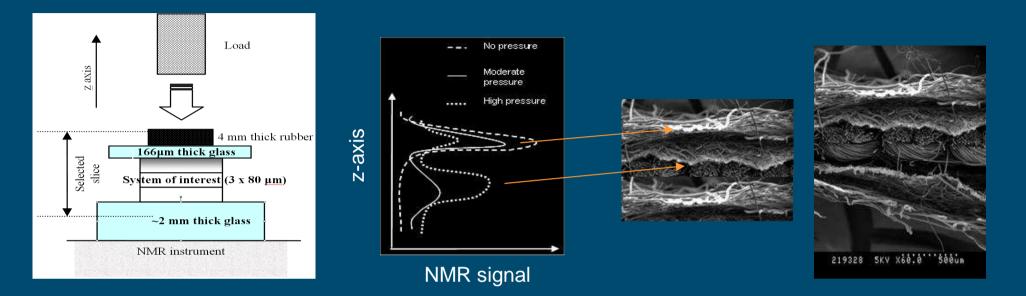


- Nuclei in the higher energy state return to the lower state is by the relaxation process
  - Spin lattice (longitudinal) relaxation
  - Spin spin (transverse) relaxation
- Spin lattice relaxation T<sub>1</sub>
- Spin spin relaxation T<sub>2</sub>
- CPMG sequence used to collect NMR data. A profile is obtained by fitting an exponential decay for each pixel to get the local T<sub>2eff</sub> and the local NMR signal amplitude.





### **Assessment of Textile Substrates**

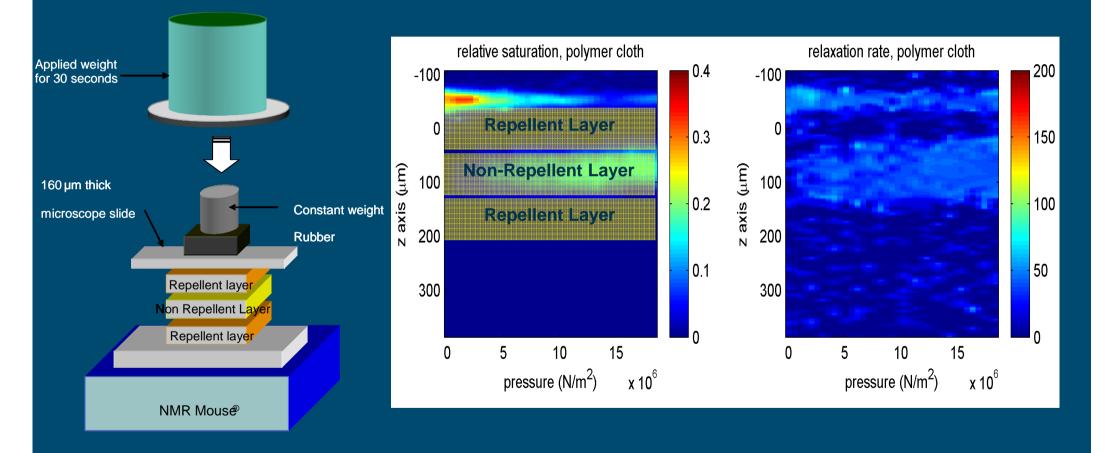


- Textiles that provide protection against toxic chemicals need to prevent the ingress of aerosols, vapours and liquids
  - Aerosols: particle capture
  - Liquid: repellent and wicking layers
  - Vapour: activated carbon
- Test methods that image a textile's performance are desirable





### Spatially resolved measurement

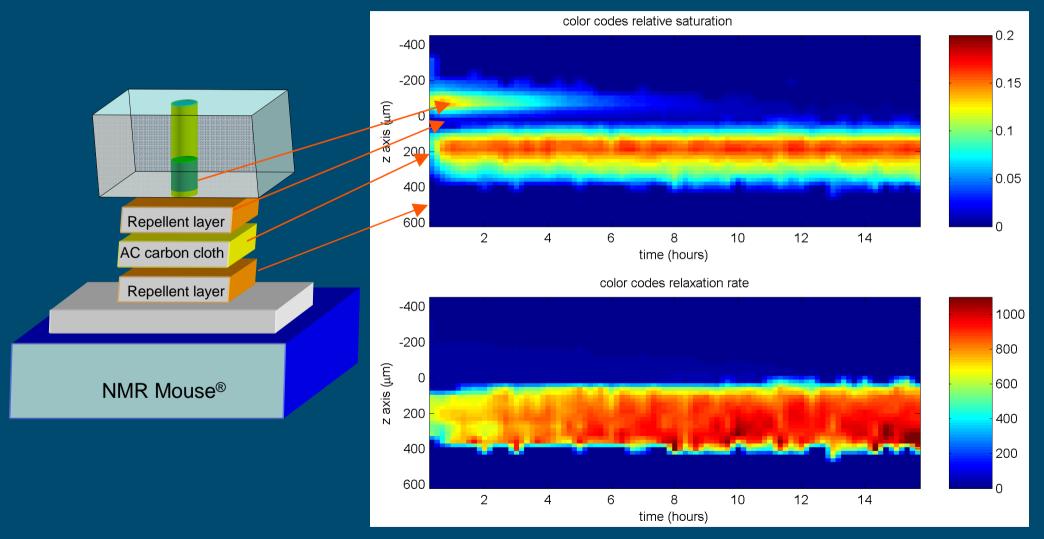






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## Spatially resolved measurement of Vapour uptake through a repellent layer





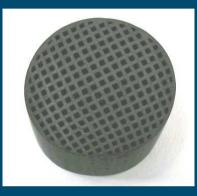


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# Different types of Activated Carbon (AC) for this study



AC cloth



AC monolith: Is made from pure carbon extruded.



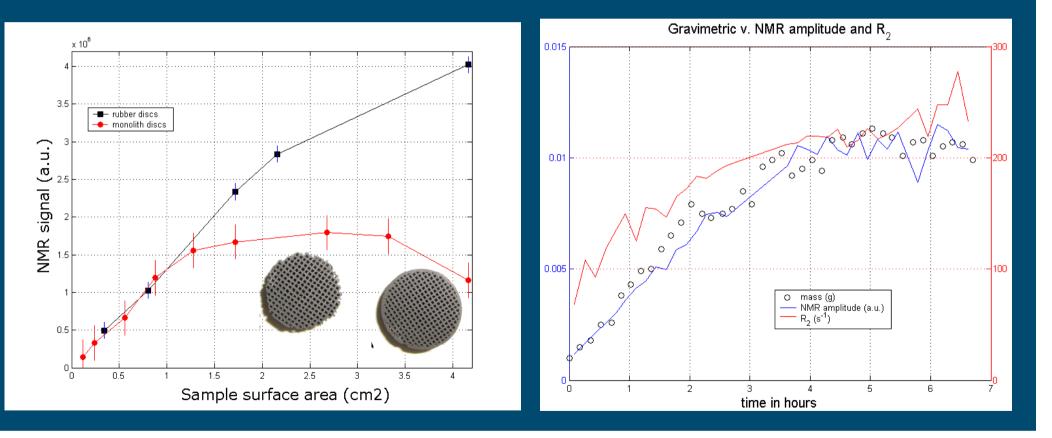
AC composite





## Conductivity and its Affect On The NMR Amplitude

- Graph showing signal from rubber discs v. monolith discs as a function of disc area
- Graph of gravimetric v. NMR amplitude for the thin sample

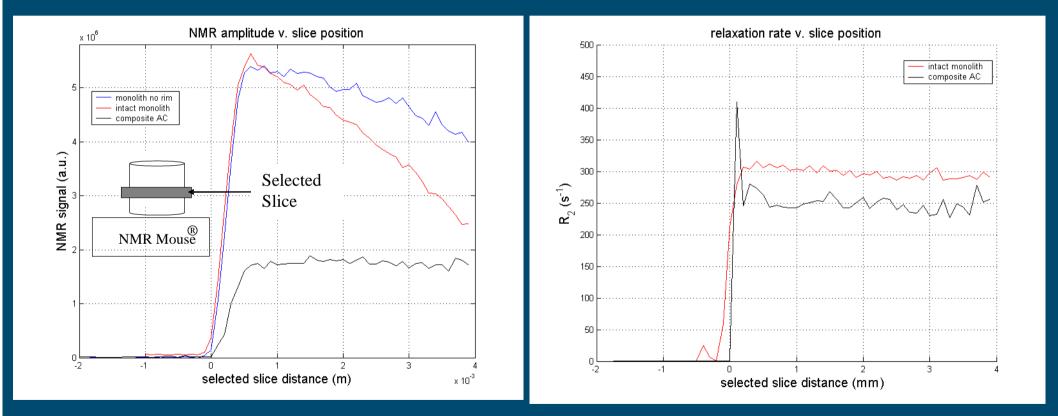






#### **Quantification of the NMR Signal Intensity as a Function of Sample Diameter**

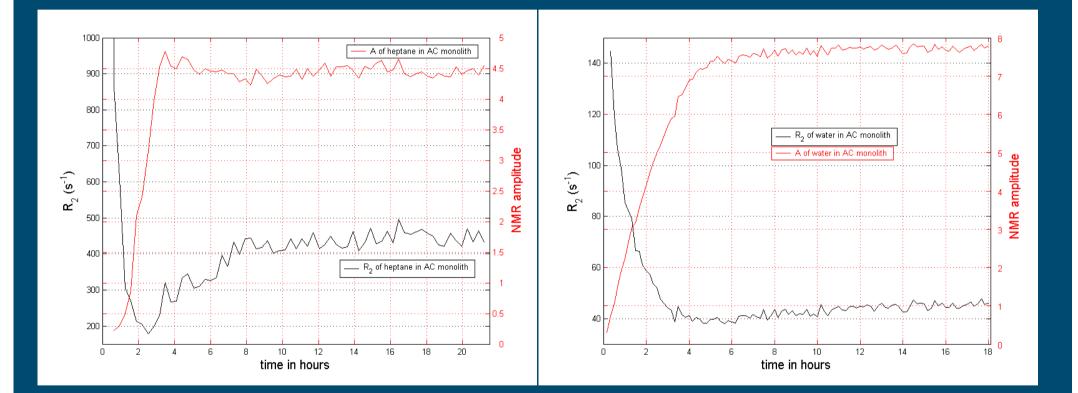
- Graph of the NMR amplitude for heptane-saturated AC, as a function of depth
- Variation of T2 with z at equilibrium







### Relaxation rate variation with Time For Water and Heptane Adsorption (pore filling?)







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### Conclusions

- Unilateral NMR is capable of imaging both liquid ingress and vapour adsorption in a multi-layered fabric.
- The instrument provides information on the quantity of liquid or vapour present in a x-y plane.
- Depth profiling (z- direction) is also achieved without the need to move the sample physically.
- In spite of the high conductivity of carbon substrates, the unilateral NMR instrument is able to monitor vapour uptake in dense activated carbons.
- The time course of the relaxation rate clearly exhibits two different regimes, a rapid and early decrease corresponding to pore filling, followed by a slower rise that we do not understand yet.
- Unilateral NMR is a simple and relatively inexpensive instrument that provides very useful insight in the uptake of molecules in a porous material under standard test regimes.





### Acknowledgements





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